Tetrahedron Letters, Vol.26, No.48, pp 5883-5886, 1985 0040-4039/85 \$3.00 + .00 Printed in Great Britain ©1985 Pergamon Press Ltd.

> Intramolecular Photocyclization of 1-(Trimethylsily1)-1,5-Hexadien-3-ones Phyllis Wilson, Steven Wolff, and William C. Agosta Laboratories of The Rockefeller University New York, New York 10021-6399

Dedicated to Professor Harry H. Wasserman on the occasion of his sixty-fifth birthday.

<u>Abstract</u>: An influence upon the regiochemistry of cyclization due to a 1-trimethylsilyl substituent is indicated upon comparison of the photochemistry of $4a_{ee}b$ and their carbon analogs $5a_{ee}b$.

In previous studies we have demonstrated the importance of structural features in controlling the regiochemistry of intramolecular photochemical cyclization of 1,5-hexadien-3-ones.¹ The parent system ($l_{,}$, see Table) undergoes 1,5-cyclization and subsequent closure to give bicyclo[2.1.1]hexan-2-one following the familiar "rule of five";² however, if the dienone is substituted with an alkyl group at C(5) (as in $l_{,}$) or the conjugated double bond is incorporated in a five- or six-membered ring ($l_{,}$), 1,6-closure becomes competitive. We now report that a trimethylsilyl (TMS) group at C(1) has an effect upon the cyclization of these dienones and that this is not a steric effect. The only relevant previous report³ known to us concerned a different sort of regiochemical effect of the trialkylsilyl group in intermolecular photocycloadditions.

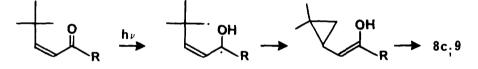
We examined ketones $4\pi_{A}p$ and $5\pi_{A}p$. These were readily prepared from <u>E</u>-3-trimethyl-



sily1-2-propenal⁴ and E-4,4-dimethy1-2-pentenal⁵ by addition of ally1- or methally1magnesium chloride and oxidation of the carbinol.^{6,7} Dilute solutions of the dienones were irradiated through an uranium glass filter (λ >340 nm) containing \sim 6% of methanol or <u>t</u>-butanol. A photostationary mixture of the <u>E</u> and <u>Z</u> isomers of 4 and 5 was rapidly established,⁸ followed by

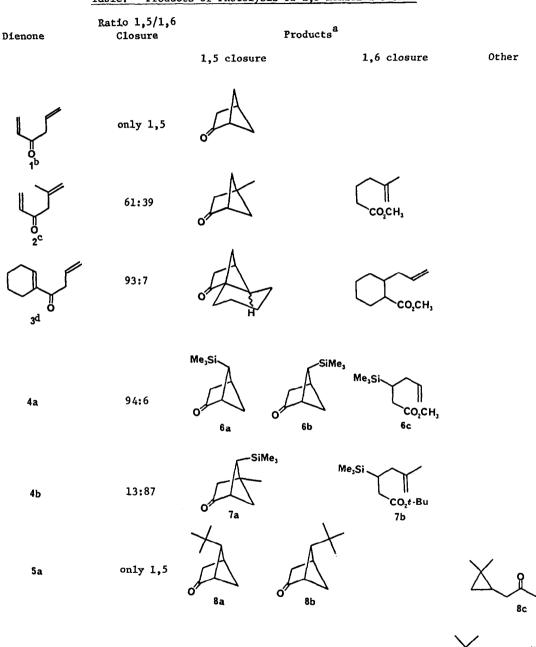
slow formation of the products (1-6 days for complete conversion). The results of these irradiations are collected in the Table.

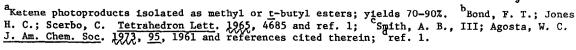
Photolysis of 4a yielded the expected¹ bicyclic ketones $6a p p^9$ and a small amount of 1,6 closure product 6c. Irradiation of 4b afforded largely ester 7b.¹¹ The C(5) substituent and ring effects noted above are roughly additive, and in molecules containing both, highly regioselective 1,6 closure can occur.¹ Apparently the C(1) TMS group also acts synergistically when a C(5) methyl group is present. Irradiation of <u>t</u>-butyl dienones 5a p p gave no 1,6-closure products and thus no unusual¹² steric effect appears to be operating in the formation of 6c and 7b. There is precedent for the formation of cyclopropyl ketones from t-butyl enones;¹³ a reasonable mechanism is shown below.



$$R = CH_2CH = CH_2; CH_2C(CH_3) = CH_2$$

The mechanism whereby the TMS group influences the cyclization of $\frac{4}{5}$ is not yet clear. Although nothing is known about the influence of silyl substitution on excited states of enones, its influence on carbon-centered radicals is well established. β -Silyl substitution imparts particular stability to radicals, ¹⁴ and this fact can explain the regioselectivity observed in [2+2] photocycloaddition of olefins to 2- and 3-trialkylsilyl-2cyclopenten-1-ones.³ Furthermore, α -silyl substitution is also known to alter radical reactivity.¹⁵ There is good evidence that isomerization of hexadienones to bicyclo[2.1.1]hexanones involves initial 1,5, rather than 2,6, bonding,¹ so that our present results suggest that the C(1) TMS group somehow favors 1,6 over 1,5 closure. The magnitude of this effect is about the same as that resulting from incorporation of the conjugated double bond in a six-membered ring, so that one interesting mechanistic speculation is that the geometric requirement for β -silyl stabilization in the transition state leading to bonding at C(1) limits torsion about the C(1)-C(2) bond roughly to the extent possible in a cyclohexene. Investigations are now underway to explore this possibility.¹⁶



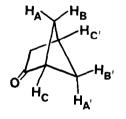


5ь

9

References

- Agosta, W. C.; Wolff, S. J. Org. Chem. 1980, 45, 3139; Wolff, S.; Agosta, W. C. 1. J. Am. Chem. Soc. 1983, 105, 1292; Matlin, A. R.; Wolff, S.; Agosta, W. C. Tetrahedron Lett. 1983, 24, 2961.
- 2. Srinivasan, R.; Carlough, K. H. J. Am. Chem. Soc. 1967, 89, 4932; Liu, R. S. H.; Hammond, G. S. J. Am. Chem. Soc. 1967, 89, 4936.
- Shih, C.; Fritzen, E. L.; Swenton, J. S. J. Org. Chem. 1980, 45, 4462. 3.
- Jung, M. E.; Gaede, B. Tetrahedron 1979, 35, 621; see also Denmark, S. E.; Jones, 4. T. K. J. Org. Chem. 1982, 47, 4595.
- 5. Lau, K. S. Y.; Schlosser, M. J. Org. Chem. 1978, 43, 1595.
- Corey, E. J.; Suggs, J. W. Tetrahedron Lett. 1975, 2647. 6.
- 7. The dienones and their products (all new compounds) were purified by VPC and gave satisfactory ¹H NMR (60- or 300-MHz), IR, and high resolution mass spectra (EI or CI) and/or elemental analyses.
- The Z isomer formed in the photolysis of 4b was isolated and characterized. 8.
- The stereochemistry of the C(5) substituent can be assigned upon examination of the 9. 300-MHz ¹H NMR of these compounds. For endo isomers 6a and 8a (H_A absent), H_A, is



observed at high field as a doublet of doublets coupled to $\begin{array}{c} H_{A} \\ H_{B} \\ H_{C} \\ H_{C} \\ H_{C} \\ H_{C} \\ H_{A'} \end{array} \qquad \begin{array}{c} H_{B}, \text{ and } H_{C} (\text{or } H_{C}). & \text{For } 6a: H_{A}, \delta 1.62, J_{A'C} = 0.7, J_{A'B'} = 0.6, J_{A'B'} = 0.7, J_{A'B'} = 0.6, J_{A'B'} = 0.6, J_{A'B'} = 0.6, J_{A'B'} = 0.7, J_{A'B'} = 0.7, J_{A'B'} = 0.6, J_{A'B'} = 0.7, J_{A'A'} = 0.7, J_{A'A'} = 0.7, J_{A'A'} = 0.7, J_{A$

for \S_{D} : 6 1.75, H_{A} , J_{AA} , = 6.5 Hz; 1.52, H_{A} , J_{AA} , = 6.7, $J_{A'B'}$, = 8.2 Hz.

- 10. Wolff, S.; Agosta, W. C. J. Org. Chem. 1980, 45, 1332 and references cited therein.
- 11. The stereospecific formation of Za will be discussed in our full paper.
- 12. No 1,6 closure products are observed in the photolysis of 1,1-dimethy1-1,5-hexadien-3-one.1
- 13. Jorgenson, M. J.; Yang, N. C. J. Am. Chem. Soc. 1963, 85, 1698.
- 14. Sakurai, H.; Hosomi, A.; Kumada, M. J. Org. Chem. 1969, 34, 1764; Kawamura, T.; Kochi, J. K. J. Am. Chem. Soc. 1972, 94, 648.
- Wilt, J. W.; Aznavoorian, P. M. J. Org. Chem. 1978, 43, 1285; Wilt, J. W. J. Am. 15. Chem. Soc. 1981, 103, 5251; Wilt, J. W.; Belmonte, F. G.; Zieske, P. A. J. Am. Chem. <u>Soc.</u> 1983, 105, 5665.
- 16. This research was supported by a grant from the National Science Foundation. (Received in USA 3 June 1985)